



Fluid Compatibility Issues - Pesticides

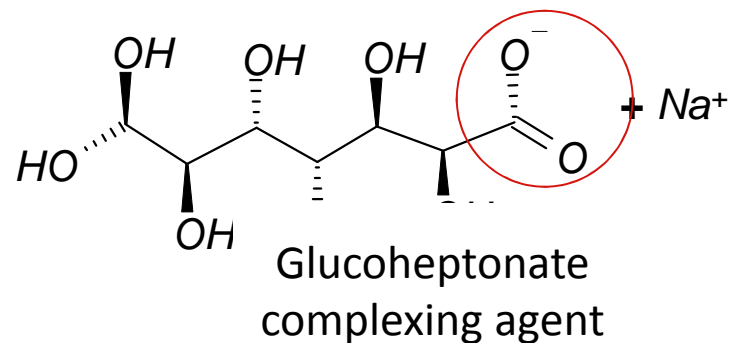
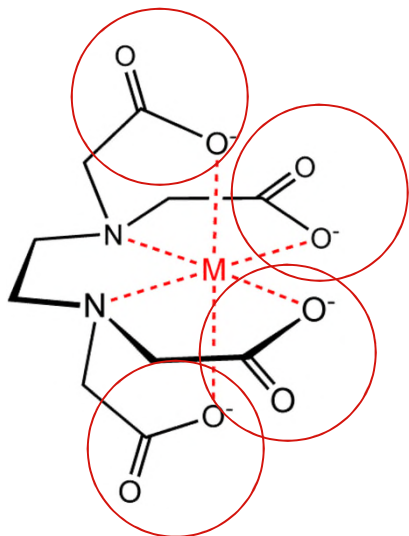
Brian Haschemeyer
December, 2013



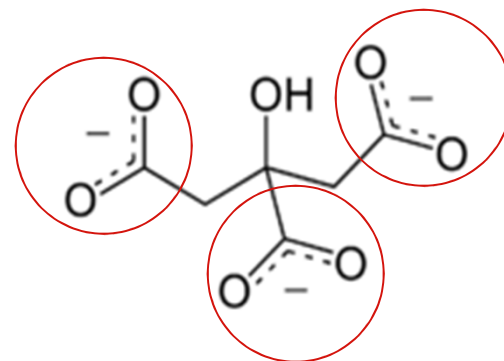
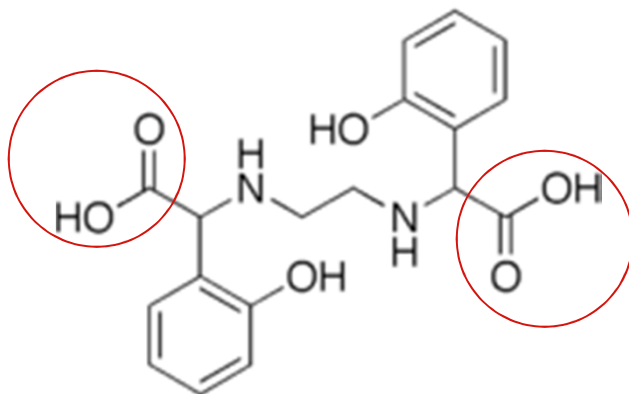
Name the Structure

What do they have in common?

EDTA
chelating
metal



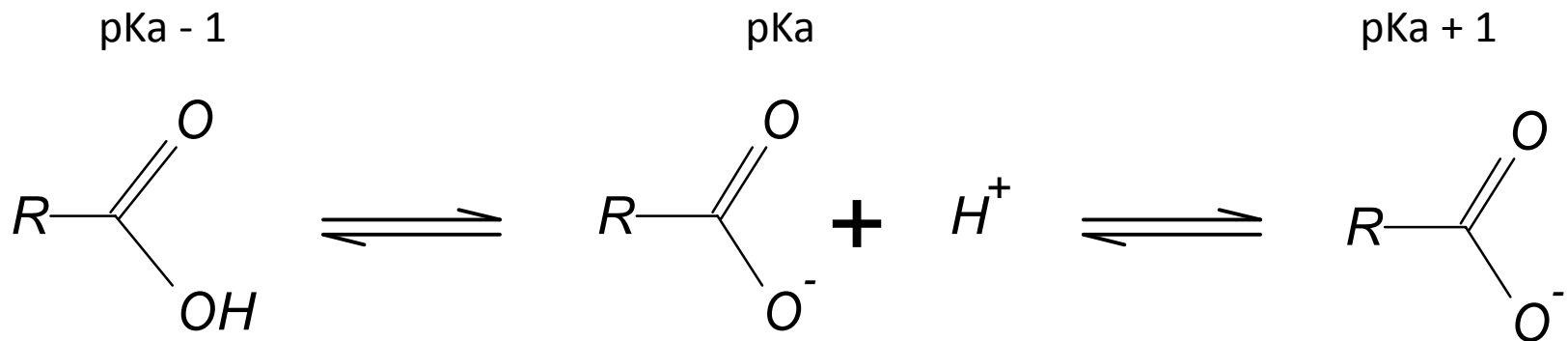
EDDHA
chelating agent



Citric Acid
chelating agent

Carboxyl group – key function group of many Ag chemicals

Chelates, Complexes and Herbicides



pKa - 1 is the pH value where the carboxyl groups exhibits no charge 100% of the time

The pKa value is pH value where the functional groups if protonated 50% of the time

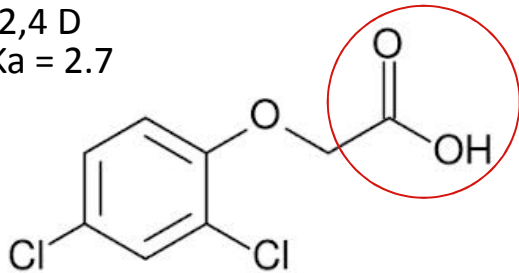
pKa + 1 is the pH value where the carboxyl groups has a negative charge 100% of the time

Phenoxy Herbicides

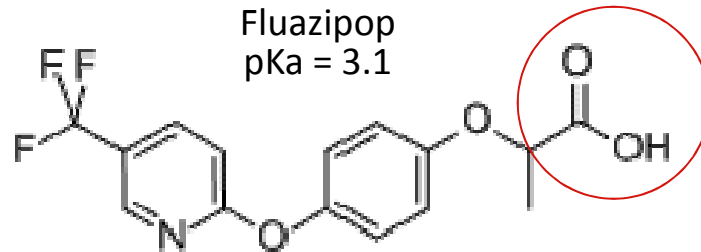
Do you recognize any functional groups?

Carboxyl groups

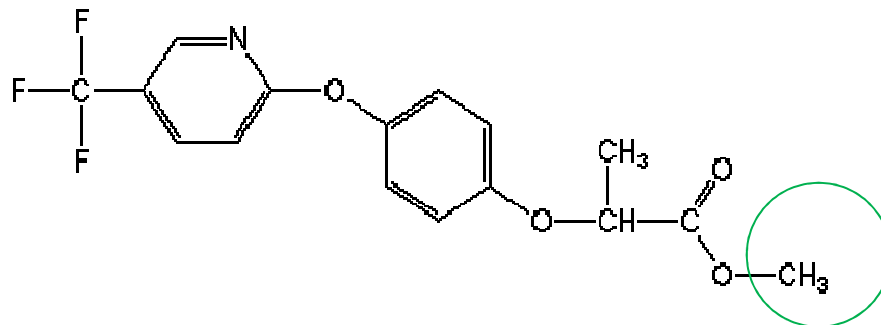
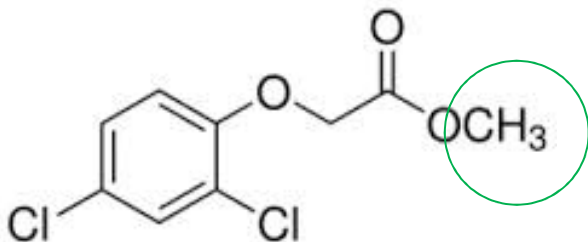
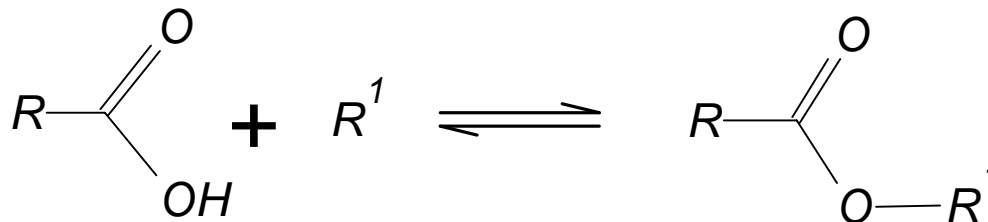
2,4 D
pKa = 2.7



Fluazipop
pKa = 3.1



What about the ester formulation?



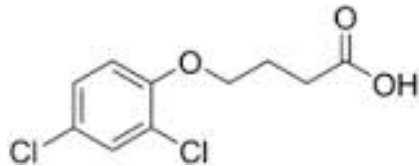
Phenoxy herbicides

Contain carboxyl groups

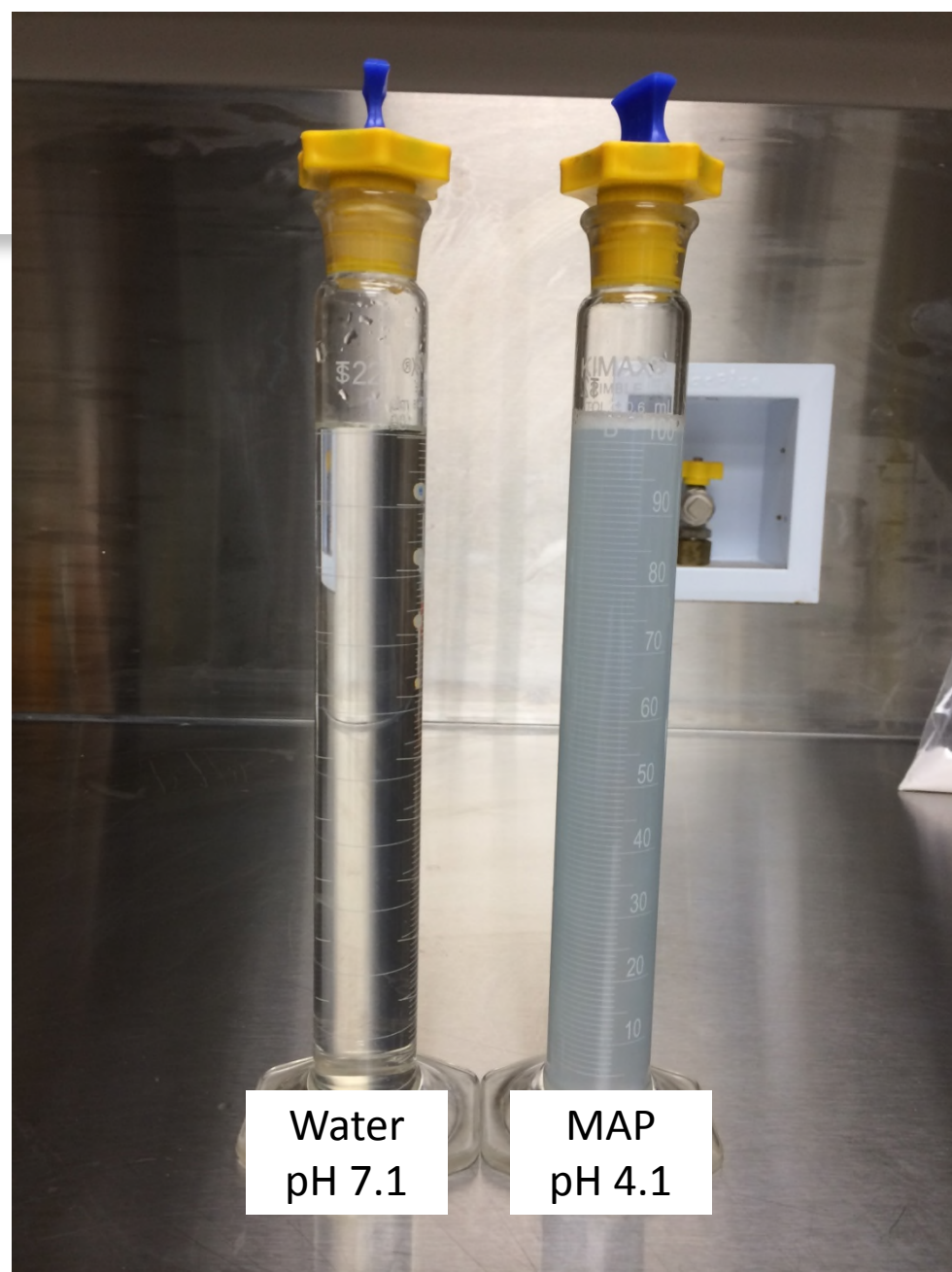
Active Salt / Ester	pKa	pka + 1	Formulation Type	Chemical Class	acid structure	Water Solubility Notes
2,4-D	2.7	3.7	SL	phenoxy acid	R ₁ -COOH	Acid and ester forms are sparingly soluble, the salts have high solubility. Formulated as both a water salt and oil soluble ester
2,4-DB	4.8	5.8	SL	phenoxy acid	R ₁ -COOH	Acid and ester forms are sparingly soluble, the salts have high solubility. Formulated as both a water salt and oil soluble ester
Fenoxaprop-P	3.2	4.2	EC	phenoxy acid	R ₁ -COOH	Sparingly soluble - Products on market are Emulsifiable Concentrates
Fluazifop	3.1	4.1	EC	phenoxy acid	R ₁ -COOH	Sparingly soluble - Products on market are Emulsifiable Concentrates
Fluazifop-P-Butyl	2.9	3.9	EC	phenoxy acid	R ₁ -COOH	Sparingly soluble - Products on market are Emulsifiable Concentrates

pH precipitation soluble liquid herbicide

- Reflex Herbicide
- Active: 2,4-DB

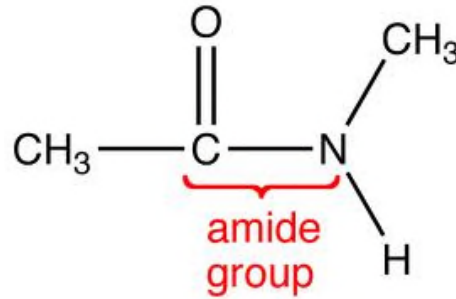
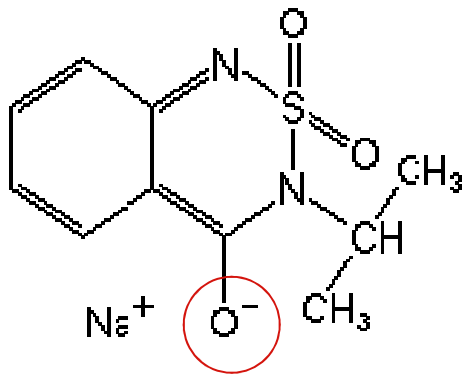


- SL Formulation
- $pK_a = 4.8$
- Acid form is sparingly soluble, the salts have high solubility.
Typically sold as a sodium salt

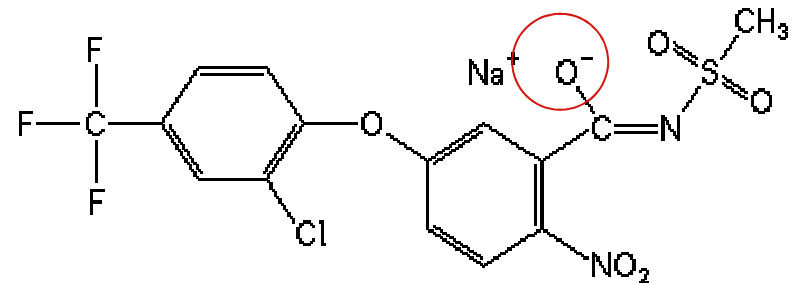


Herbicides – Amide groups

Sodium salt of Bentazon
pKa = 4.3

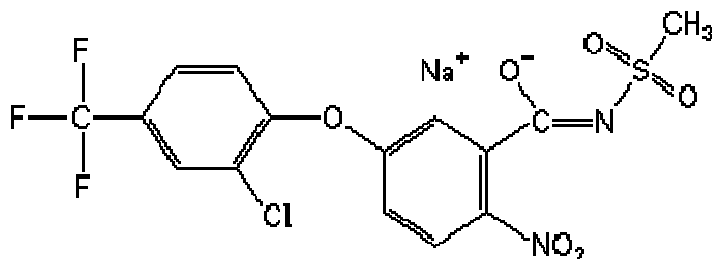


Sodium salt of Fomesafen
pKa = 3.8

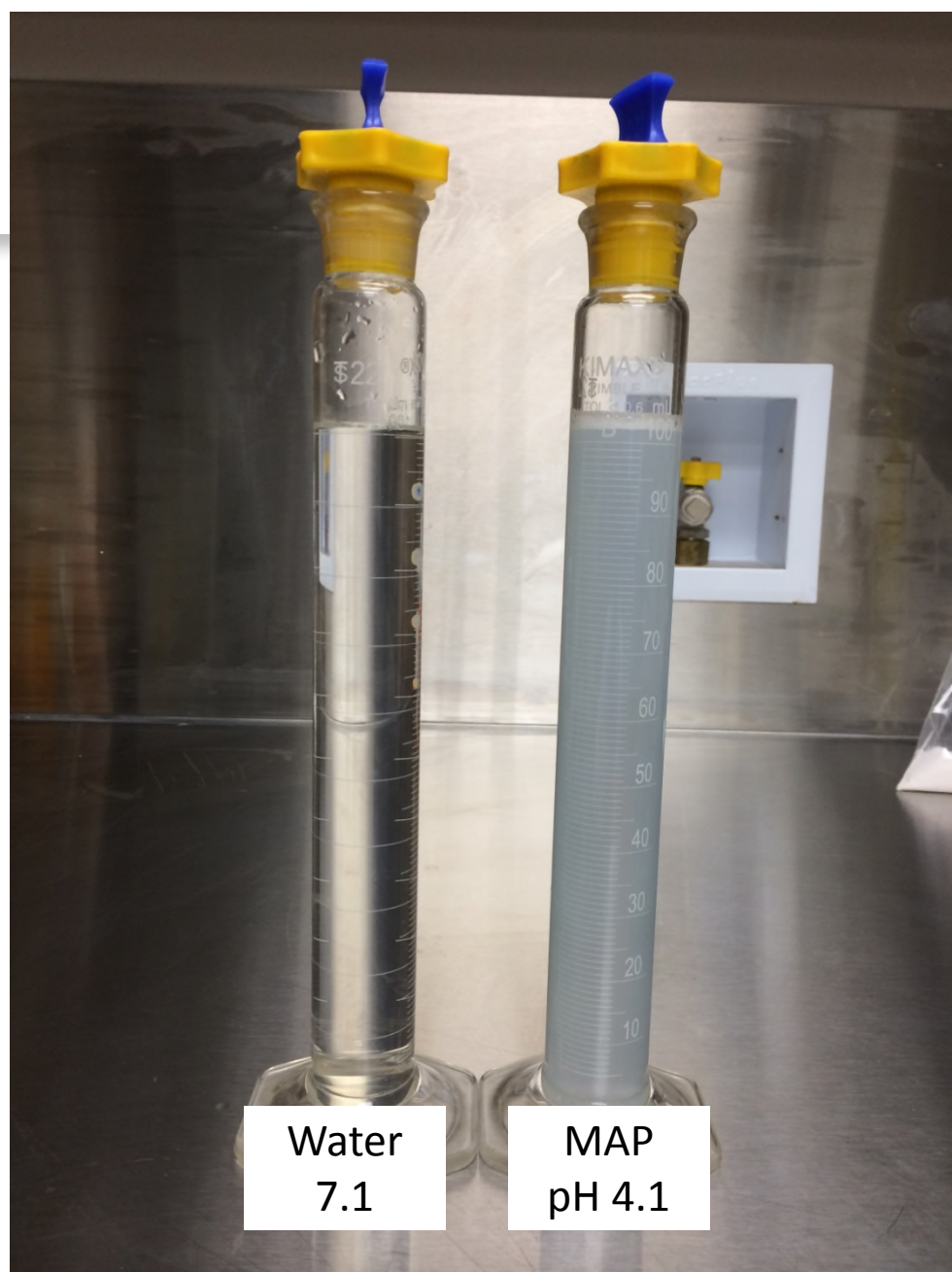


pH precipitation soluble liquid herbicide

- Reflex Herbicide
- Active: Sodium Fomesafen

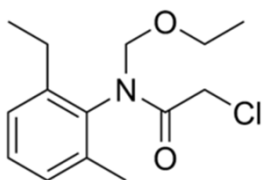


- SL Formulation
- pKa = 3.8
- Acid form is sparingly soluble, the salts have high solubility.
Typically sold as a sodium salt

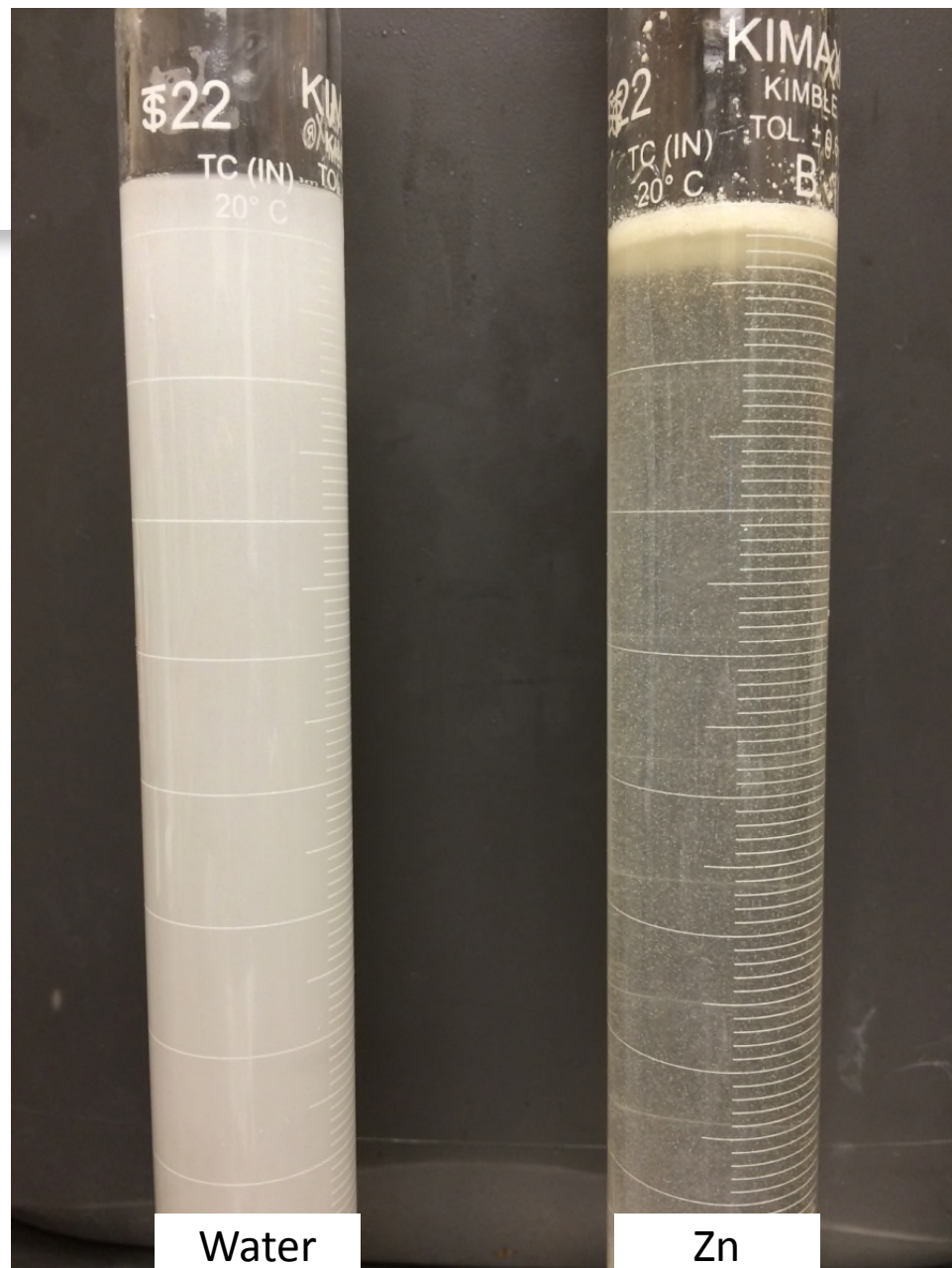


Suspension concentrate failure in presence of divalent cations

- Warrant Herbicide
- Active: Acetochlor

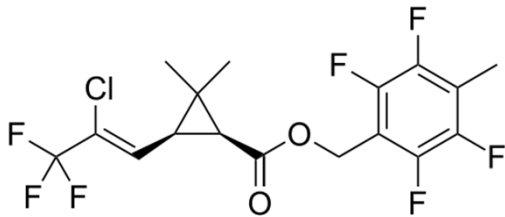


- SC Formulation
- Anionic dispersant fails do to binding divalent cations binding to the negative charged sites of the dispersant.

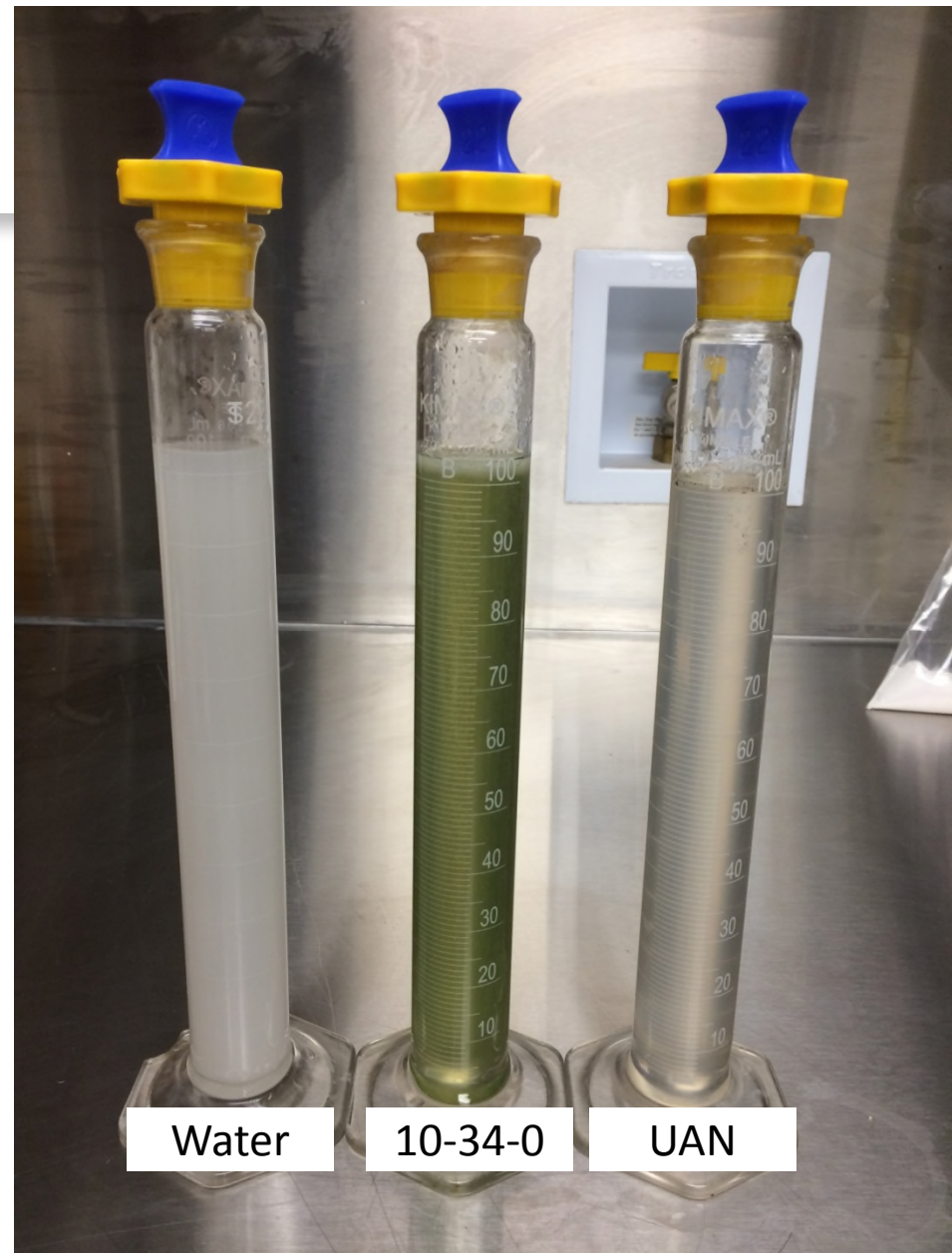


Suspension Concentrate failure in 10-34-0

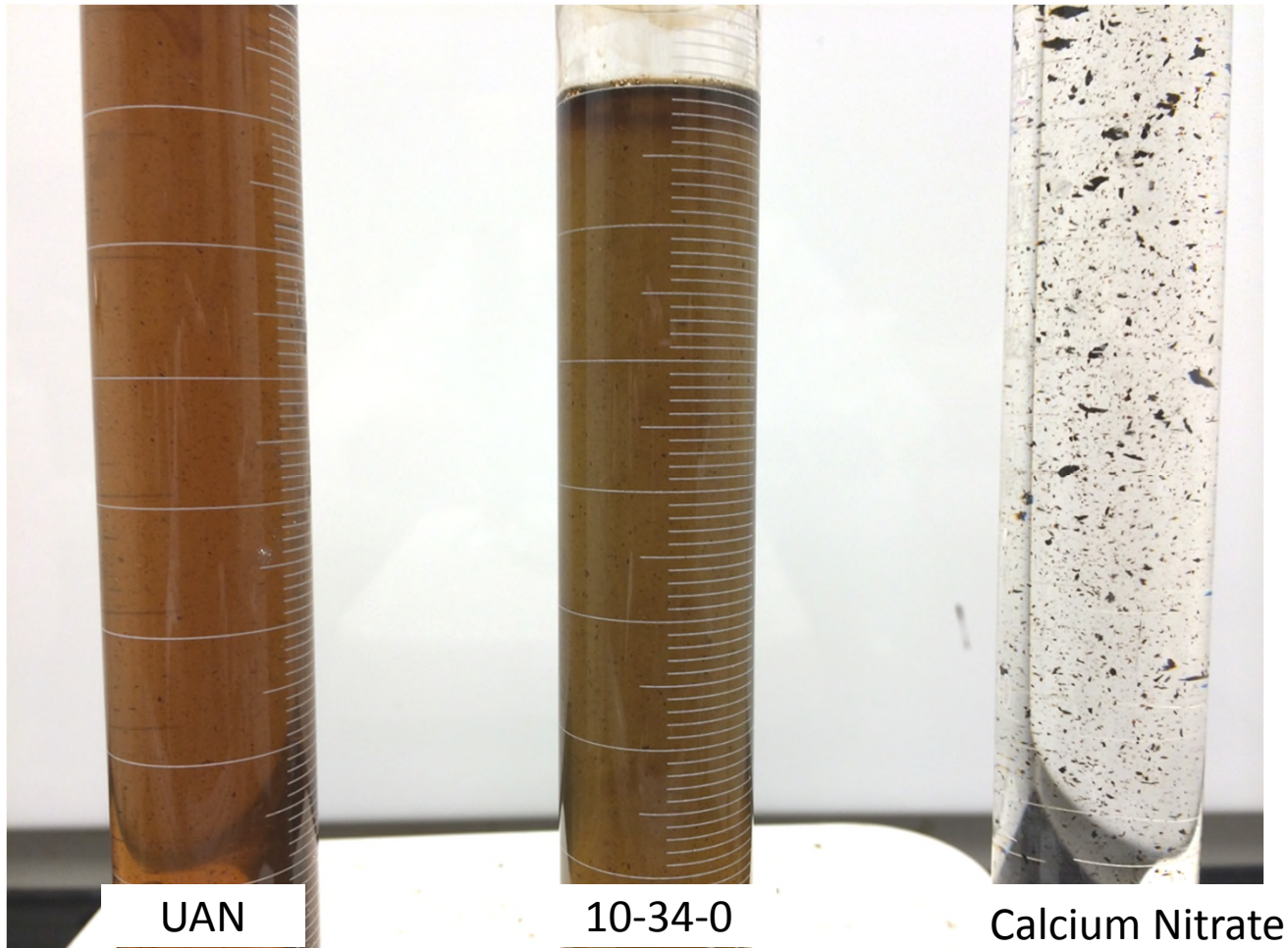
- Force Insecticide
- Active: Tefluthrin



- SC Formulation
- Sparingly soluble in water, liquid formulations are typically SC or EC
- Dispersant fails do to limited water to activate dispersing and emulsifying agents.



Humic Acid in Liquid Fertilizer



UAN and ATS with Pre-emerge Herbicide

- Lexar EZ, Bicep II Magnum, etc.
- SC Formulation
- Sparingly soluble in water, liquid formulations are typically SC or EC
- Dispersant fails do to limited water to activate dispersing and emulsifying agents. Hi electrolyte solution, limited free water.



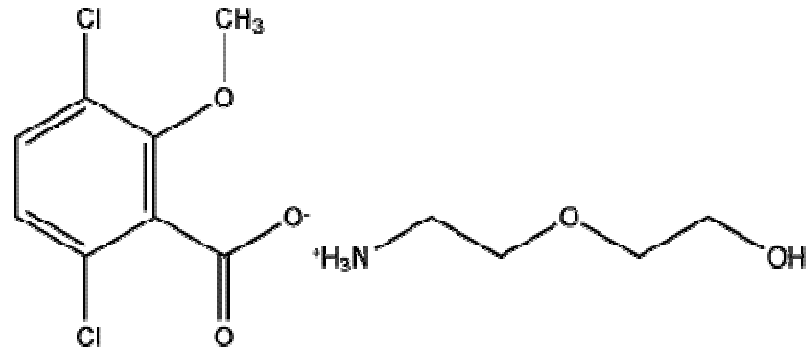
Counter Ion Affects Volatility of Dicamba

Ammonium can increase volatility

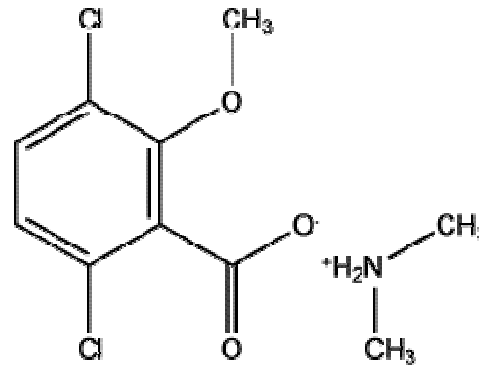
- Be caution of adding ammonium containing liquid fertilizers with Dicamba



- AMS Solutions
- ATS Solutions
- UAN Solutions
- MAP Solutions



Dicamba, Diglycolomine salt

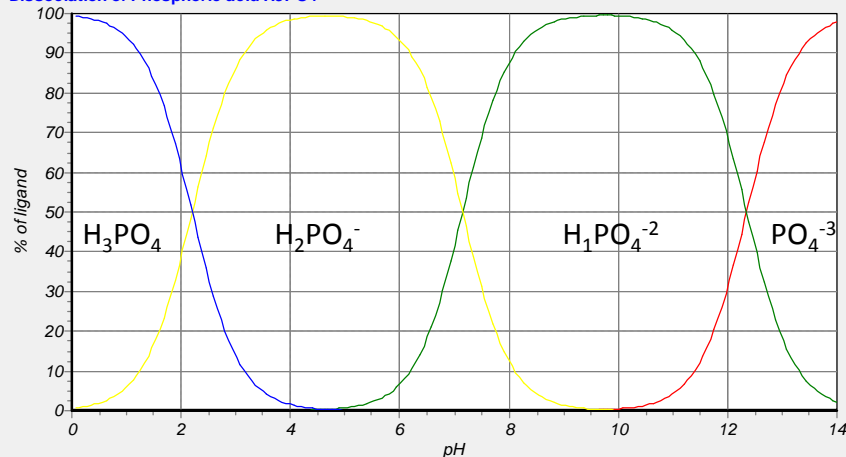


Dicamba, Dimethylamine salt

Phosphate Interactions

pH Dependence of Phosphate Binding

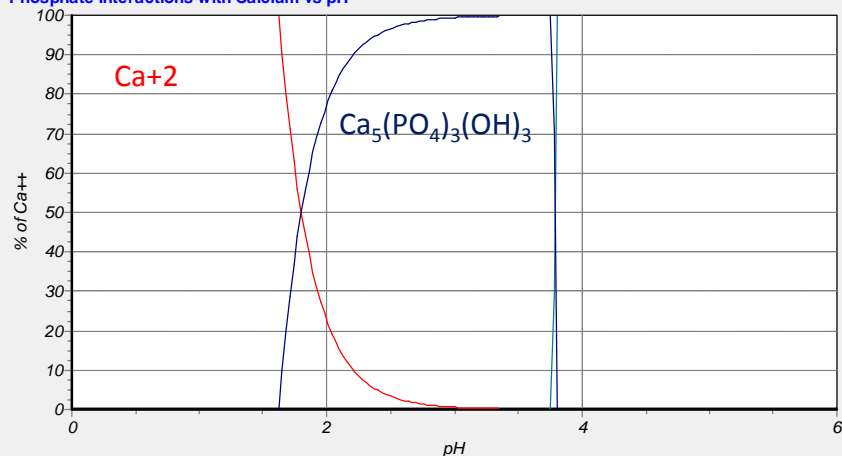
Dissociation of Phosphoric acid H₃PO₄



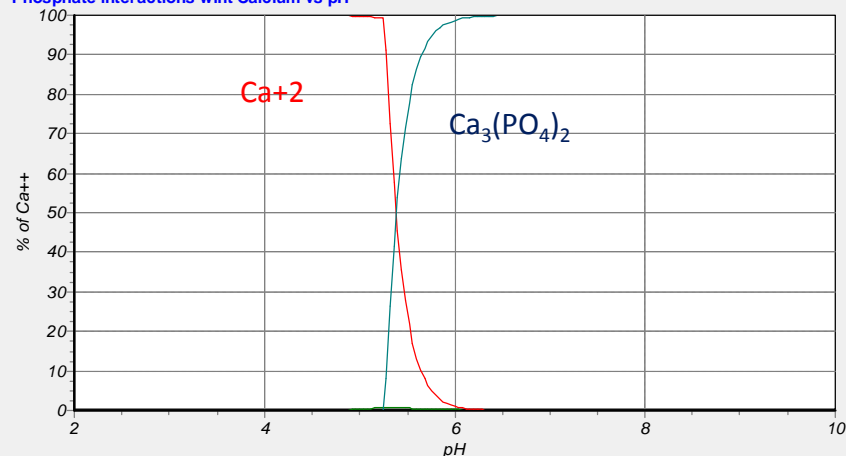
Acid	Mol. Form	pKa
H ₃ PO ₄	H ₂ PO ₄ ⁻	2.2
	H ₁ PO ₄ ⁻²	7.2
	PO ₄ ⁻³	12.3

[M] ⁺ⁿ	Form	Ksp
Ca+2	Ca ₃ (PO ₄) ₂	1x10 ⁻²⁶
	Ca ₂ (PO ₄) ₂ (OH) ₂	1x10 ⁻²⁷
	Ca ₅ (PO ₄) ₃ (OH) ₃	1x10 ⁻⁵⁷

Phosphate Interactions with Calcium vs pH



Phosphate interactions with Calcium vs pH

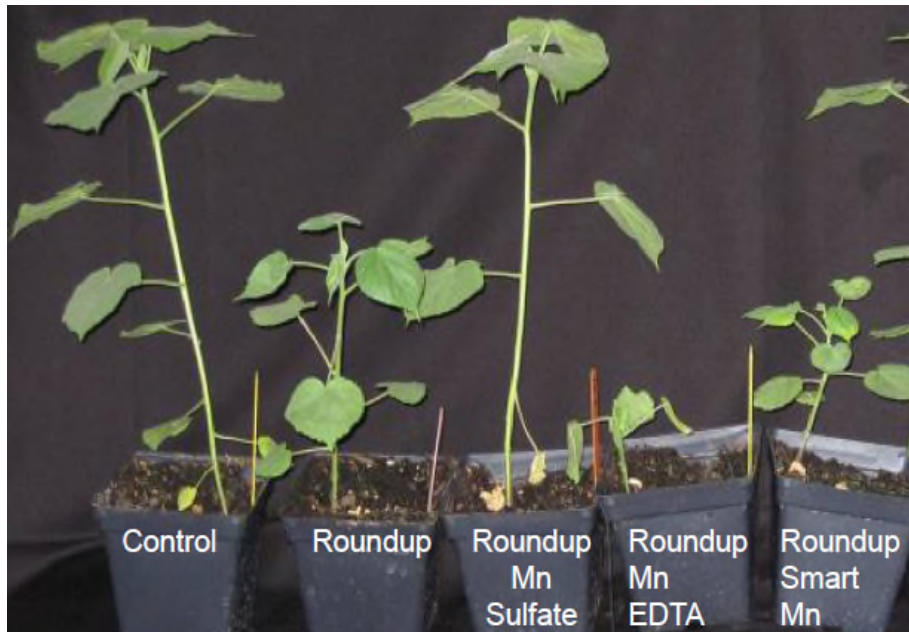
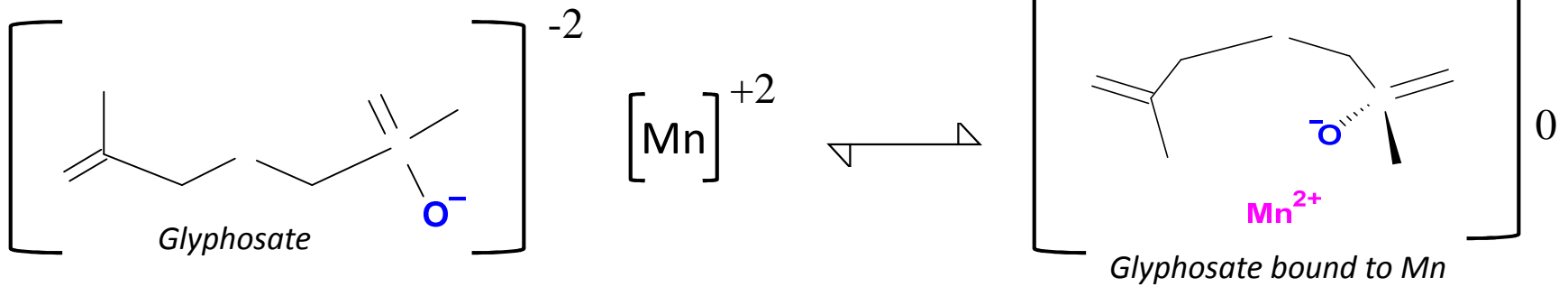


Graph based on 200mmol phosphate & 20mmol Ca+2 concentrations

BRANDT

Glyphosate

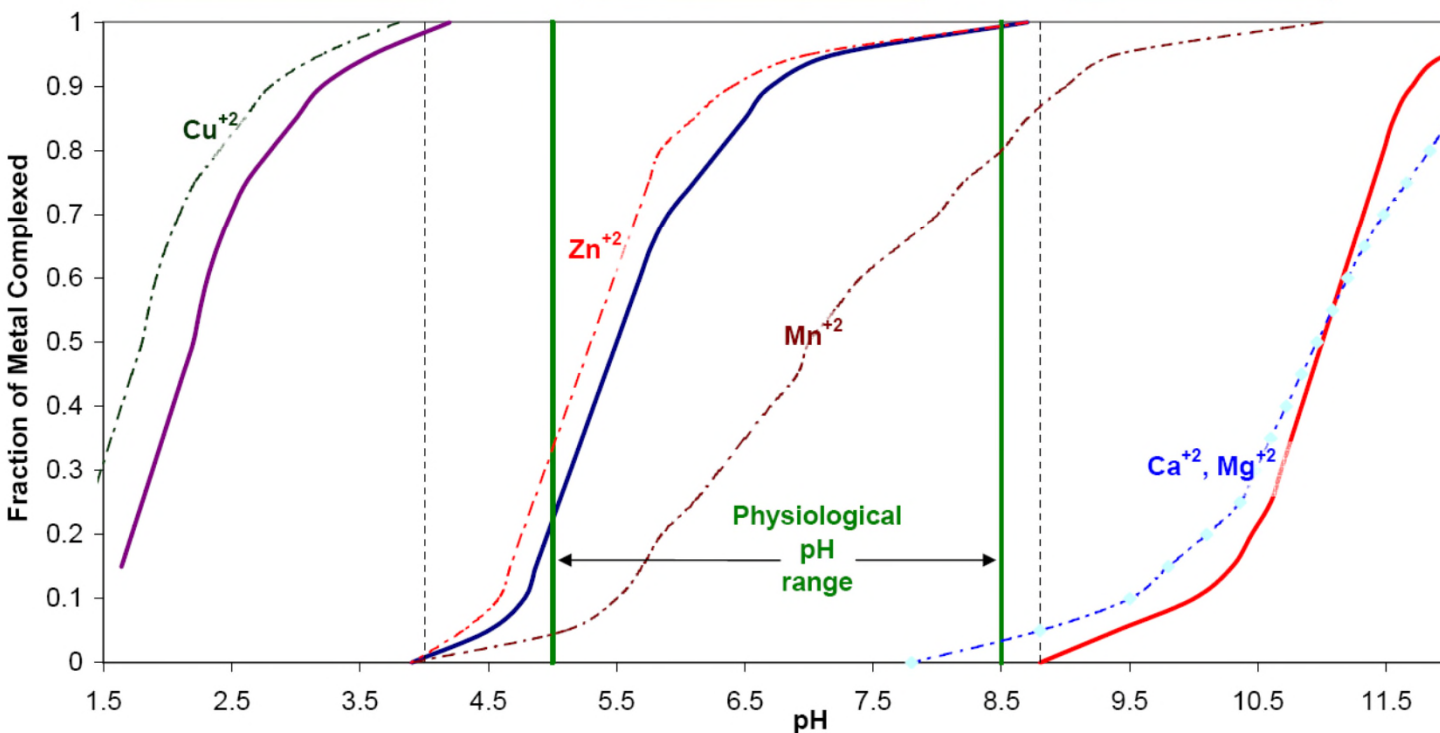
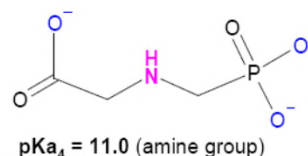
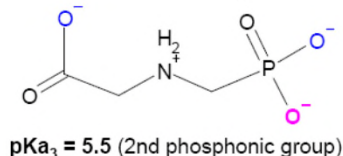
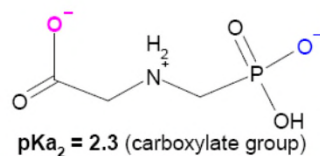
Antagonized by Divalent Cations



Glyphosate – Solution Chemistry

Interactions between Cations and Glyphosate

Metal Complexes in Relation to Dissociation of Glyphosate vs Solution pH



Stability Constants

1:1 molar ratio

@ physiological pH

Cation	(Log K_{m1})
Ca⁺²	3.3
Mg⁺²	3.3
Cu⁺²	11.2
Fe ⁺²	6.9
Fe ⁺³	16.1
Mn⁺²	5.5
Zn⁺²	8.4

Foliar Micronutrients

Herbicide Safener – Stress Mitigation

Alfalfa - 2012
Pacific Northwest



Safening affect observed with
Brandt Smart Trio (Zinc and
Manganese) @1 qt / acre

- Butyrac / 2,4 DB
- Bromoxynil
- Nitrogen Based Surfactant
- 20 gal of water/acre



Foliar Micronutrients
Herbicide Safener – Stress Mitigation

Cotton - 2013
Virginia



Foliar Micronutrients Herbicide Safener – Stress Mitigation

Sugarcane
Brazil

Gesapax (Ametrina) 3 litros/ Hectare
DMA (2-4D) 0,4 litros/ Hectare
Ancosar 720 (MSMA) 2,2 litros/ Hectare



Brandt Smart Trio® 2 liter/Ha
Gesapax (Ametrina) 3 litros/ Hectare
DMA (2-4D) 0,4 litros/ Hectare
Ancosar 720 (MSMA) 2,2 litros/ Hectare



Thank You

- Jar Test

